

## Numericals Chemistry Chapter Solid State

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### Numericals Chemistry Chapter Solid State

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Solid State Chemistry Numericals The post-Solid State Chemistry Numericals contain all important questions related to the topic Que 1. A solid has a cubic structure in which X atoms are located at the corners of the cube, Y atoms are at the cube centers and O atoms are at the edge

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### Numericals Chemistry Chapter Solid State

Numericals Chemistry Chapter Solid State Solid state numerical 1.14 If the radius of the octahedral void is  $r$  and radius of the atoms in closepacking is  $R$ , derive relation between  $r$  and  $R$ . Lecture i... Solid State Numerical - 1.14 | Chapter - 01, Class - 12 ...

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What type of solid is it? Ans: It is a covalent or network solid. 1.8. Why are ionic solids conducting in the molten state and not in the solid-state? Ans: In the ionic solids, the electrical conductivity is due to the movement of the ions. Since the ionic mobility is negligible in the solid state, these are non-conducting in this state.

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Solids have definite volume, shape, and mass due to the short distance between the fixed position of particles. To learn more access Solid State Chemistry Class 12 Notes PDF on BYJU'S.

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Chemistry Notes for class 12 Chapter 1 The Solid State Solids Solids are the chemical substances which are characterised by define shape and volume, rigidity, high density, low compressibility. The constituent particles (atoms, molecules or ions) are closely packed and held together by strong interparticle forces Types of Solids

### Chemistry Notes for class 12 Chapter 1 The Solid State

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