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left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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The number  $6.022 \times 10^{23}$ , called Avogadro's number after the 19th-century chemist Amedeo Avogadro, is the number we use in chemistry to represent macroscopic amounts of atoms and molecules. Thus, if we have  $6.022 \times 10^{23}$  Oxygen atoms, we say we have 1 mol of Oxygen atoms.

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following questions in the space  
provided. 1. \_\_\_\_\_ When an electron is  
added to a neutral atom, energy is (a)  
always absorbed.



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