

# File Type PDF Definition Of Hypertonic Solution

## Definition Of Hypertonic Solution

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## **Definition Of Hypertonic Solution**

Hypertonic Solution Definition. A

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hypertonic solution contains a higher concentration of solutes compared to another solution. The opposite solution with a lower concentration is known as the hypotonic solution. Scientists must describe cell contents compared to the environment.

## **Hypertonic Solution - Definition and**

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## **Examples | Biology ...**

A hypertonic solution is one which has a higher solute concentration than another solution. An example of a hypertonic solution is the interior of a red blood cell compared with the solute concentration of fresh water.

## **What Is a Hypertonic Solution? -**

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## **ThoughtCo**

hypertonic solution one having an osmotic pressure greater than that of a standard of reference. hypobaric solution one having a specific gravity less than that of a standard of reference. hypotonic solution one having an osmotic pressure less than that of a standard of reference.

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## **Hypertonic solution | definition of hypertonic solution by ...**

Hypertonic solution: A solution that contains more dissolved particles (such as salt and other electrolytes) than is found in normal cells and blood. For example, hypertonic solutions are used for soaking wounds.



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## **Definition of Hypertonic solution - MedicineNet**

A hypertonic solution may be any solution that has a greater concentration of solutes outside of the cell, and one example of this is saline solutions used in medical care.

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## **Hypertonic Solution: Definition, Effect & Example - Video ...**

A hypertonic solution is a particular type of solution that has a greater concentration of solutes on the outside of a cell when compared with the inside of a cell. Common Hypertonic Solutions. Here are a few examples: Saline solution, or a solution that contains salts,

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is hypertonic. This type of hypertonic solution is extremely common.

## **Hypertonic Solution Examples - YourDictionary.com**

So when we say that an IV solution is Hypertonic, what we are really saying is that the hypertonic fluid has a higher solute to solvent ratio than blood does.

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For the purposes of IV fluid, this means that blood becomes our definition of what " Isotonic " means.

## **Hypertonic Solution: An Explanation for Nursing Students ...**

Hypertonic Solution A hypertonic solution is a solution that contains more solute than the cell which is placed in it.

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If a cell with a NaCl concentration of 0.9% is placed in a solution of water with a 10% concentration of NaCl, the solution is said to be hypertonic.

### **Isotonic, Hypertonic, and Hypotonic Solutions**

It has the same concentration of solute, and so you have no net inflow.

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Hypotonic solution, you have water molecules going into the cell, the cell expanding, kind of like a filling balloon.  
Isotonic solution, no net flow.

## **Hypotonic, isotonic, and hypertonic solutions (tonicity ...**

When hypertonic solutions are used (very cautiously....most likely to be

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given in the ICU due to quickly arising side effects of pulmonary edema/fluid over load). In addition, it is preferred to give hypertonic solutions via a central line due to the hypertonic solution being vesicant on the veins and the risk of infiltration.

### **Isotonic, Hypotonic & Hypertonic IV**

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## **Fluid Solution**

Hypotonic solution: A solution that contains fewer dissolved particles (such as salt and other electrolytes) than is found in normal cells and blood.

Hypotonic solutions are commonly used to give fluids intravenously to hospitalized patients in order to treat or avoid dehydration.



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## **Definition of Hypotonic solution - MedicineNet**

adjective. Physiology. of or relating to hypertonia. Physical Chemistry. noting a solution of higher osmotic pressure than another solution with which it is compared (opposed to hypotonic ).Compare isotonic (def 1).

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## **Hypertonic | Definition of Hypertonic at Dictionary.com**

A cell whose cytosol is an extremely hypotonic solution compared to its environment will shrivel up, and is said to be plasmolyzed. This is almost always a bad state for cells, which need water for many chemical reactions. In the

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opposite situation, the environment can be a hypotonic solution compared to the cell.

## **Hypotonic Solution - Definition & Examples | Biology ...**

A hypertonic solution (pronounced "HĪ-per-TAWN-ik") is a solution with a higher concentration of solute (dissolved

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substance) than some other, specified solution (and which therefore has a higher osmotic pressure than the other solution). The solution with the lower concentration is then termed hypotonic.

## **Hypertonic solution | Online Biology Dictionary**

Hypertonic, isotonic, and hypotonic

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solutions and their effect on cells. Osmosis and tonicity. Hypertonic, isotonic, and hypotonic solutions and their effect on cells. If you're seeing this message, it means we're having trouble loading external resources on our website.

**Tonicity: hypertonic, isotonic &**

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## **hypotonic solutions ...**

Hypotonic solution. Hypotonic solution is one of two solutions that has a lower concentration of a solute. < Solute, solvent, solution definition with examples Solubility rules >.

## **Isotonic hypertonic hypotonic solutions and definition in ...**

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Hypertonic and hypotonic solution both effects the cell by changing its structural configuration. In a hypertonic solution, the cell shrinksbecause of the high concentration of water inside the cell. Therefore, water will move out from the cell into its surrounding to maintain the equilibrium both outside and inside of the cell.

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## **Differences Between Hypertonic and Hypotonic solution ...**

A hypertonic solution has a greater concentration of solutes than another solution.



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