

Read Book Periodic Table  
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Period \_\_\_\_\_ Directions: Use your Periodic table to complete the worksheet. 1.

What is the atomic symbol for silver? 2.

What is the atomic mass of mercury? 3.

Ni is the symbol for what element? 4.

The element that has the atomic number 17 is? 5. List the symbols for two

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transition metals. 6.

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The periodic table is an organized arrangement of the chemical elements,



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in order of their atomic number (number of protons), electron configurations, and their chemical properties. This ordering segregates elements according to their periodic trends that are elements with similar behavior in the same column.  
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Worksheets The idea behind these worksheets is to use the element letters to make words. The first few worksheets give you the element numbers, so all you have to do is look on the table to find the corresponding letters. The last two worksheets give you the letters only - and you need to find out the element

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numbers instead.

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Directions: Use your Periodic table to

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complete the worksheet. 1. What is the atomic symbol for silver? 2. What is the atomic mass of mercury? 3. Ni is the symbol for what element? 4. The element that has the atomic number 17 is? 5. List the symbols for two transition metals. 6. Cu, Ag, and Au are all in what group # 7. Name two noble gases . 8.

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nonmetal more family group left periodic  
table metal metalloids period right  
properties atomic number 1. The chart  
that lists the elements in an organized  
way is called the \_\_\_\_\_. 2. The periodic  
table lists the elements according to  
\_\_\_\_\_. 3.

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periodic table and write the letters of the elements that match the numbers given.

95 68 53 20AmErICa 16 91 53 7 85 2 7  
16 84 57 60 10 90 68 57 60 16 74 13 99  
9 88 7 58 6 1 53 11 18 32 7 22 11 35 53  
73 53 7

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## Periodic Trends Worksheet Answers

Page 1: 1. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, and potassium. O, C, Al, K 2. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, and aluminum. Ne, Al, S, O 3. Why does fluorine have a higher ionization energy than iodine?

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