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Holt Modern Chemistry Chapter 7 The chemical formula for an ionic compound represents one formula unit - the simplest ratio of the compound's positive ions (cations) and its negative ions (anions). Monatomic cations are identified simply by the element's name.

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formula. subscript of each element
reveals number of atoms contained in a
single molecule of a certain compound.
monoatomic ions. ions formed from a
single atom; remember to change
nonmetals to -ide ending. oxyanions.

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Chapter 7 Section 3-4 Vocabulary.
STUDY. PLAY. Formula mass. The
formula unit of a molecule or ion is the
sum of the average atomic masses of all
atoms represented in its formula.
Percentage composition.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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7H 5(NO 2) 3. 4 elements a. How many elements make up this compound? 6 oxygen atoms b. How many oxygen atoms are present in one molecule of C 7H 5(NO 2) 3? 21 atoms c. How many atoms in total are present in one molecule of C 7H 5(NO 2) 3? 4.2 1024 atoms d. How many atoms are present in a sample of 2.0 1023 molecules of C 7H 5(NO 2) 3? 4.

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CHAPTER 7 HOMEWORK 7-3

(pp.211-213) VOCABULARY Write the prefix that corresponds to each number.

Try not to use the book. Prefix Number

Number Prefix 1 6 2 7 3 8 4 9 5 10

GRAPHIC ORGANIZER Complete the chart by supplying the missing information.

Modern Chemistry • CHAPTER 7

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED

REVIEW SHORT ANSWER Answer the

following questions in the space

provided. 1. Write formulas for the

following compounds: CuCO₃ a.

copper(II) carbonate Na₂SO₃ b. sodium

sulfite (NH₄)₃PO₄ c. ammonium

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phosphate SnS₂ d. tin(IV) sulfide HNO₂
e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

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